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## **Molecular Dynamics simulations, challenges and opportunities: a biologist's prospective**

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### **Abstract**

Molecular dynamics (MD) is a computational technique which is used to study biomolecules in virtual environment. Each of the constituent atoms represents a particle and hence the biomolecule embodies a multi-particle mechanical system analyzed within a simulation box during MD analysis. The potential energies of the atoms are explained by a mathematical expression consisting of different forces and space parameters. There are various software and force fields that have been developed for MD studies of the biomolecules. MD analysis has unravelled the various biological mechanisms (protein folding/unfolding, protein-small molecule interactions, protein-protein interactions, DNA/RNA-protein interactions, proteins embedded in membrane, lipid-lipid interactions, drug transport etc.) operating at the atomic and molecular levels. However, there are still some parameters including torsions in amino acids, carbohydrates (whose structure is extended and not well defined like that of proteins) and single stranded nucleic acids for which the force fields need further improvement, although there are several workers putting in constant efforts in these directions. The existing force fields are not efficient for studying the crowded environment inside the cells, since these interactions involve multiple

32 factors in real time. Therefore, the improved force fields may provide the opportunities for their  
33 wider applications on the complex biosystems in diverse cellular conditions. In conclusion, the  
34 intervention of MD in the basic sciences involving interdisciplinary approaches will be helpful  
35 for understanding many fundamental biological and physiological processes at the molecular  
36 levels that may be further applied in various fields including biotechnology, fisheries, sustainable  
37 agriculture and biomedical research.

38

39 **Key words:** Biomolecules; Force field; Ligand; Interactions; Molecular dynamics; Molecular  
40 mechanics

41

42 **Abbreviations:** AMOEBA (Atomic Multipole Optimized Energetics for Biomolecular  
43 Simulation), Correction energy map (CMAP), cyclic AMP (cAMP), Double Electron Electron  
44 Resonance (DEER), dummy spin-labels (DS), Glutamine-binding protein (GlnBP), Kinase-  
45 inducible domain (KID), Major histocompatibility complex (MHC), Molecular dynamics (MD),  
46 Molecular mechanics (MM), Post-translational modifications (PTM), Quantum mechanics (QM),  
47 Residue specific force field1 (RSFF1), Root mean square deviations (RMSD), T cell receptors  
48 (TCRs), Visual Molecular Dynamics (VMD).

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63	<b>Index of contents</b>
64	<b>1. Introduction</b>
65	<b>2. Different force fields for studying Biomolecules</b>
66	<b>3. Points to be taken care of before embarking on a molecular dynamics experiment</b>
67	<b>4. Molecular dynamics of protein-ligand systems</b>
68	<b>4.1. Protein-small molecule complex MD analysis</b>
69	<b>4.2. Protein-protein interaction specificity and binding</b>
70	<b>4.3. Behaviour of nucleic acid-binding proteins</b>
71	<b>4.4. Post-translational modifications of proteins</b>
72	<b>5. MD analysis pertaining to lipid bilayer biomembranes</b>
73	<b>5.1. Physiological properties of plasma membrane analyzed by MD</b>
74	<b>5.2. Conformational changes and functional mechanism of the integral</b>
75	<b>membrane proteins</b>
76	<b>6. Molecular dynamics of carbohydrates</b>
77	<b>7. Molecular dynamics of DNA/RNA</b>
78	<b>8. Limitations of current force fields</b>
79	<b>8.1. Poor description of secondary structure of biomolecules by current force</b>
80	<b>fields</b>
81	<b>8.2. Available force fields are deficient to study crowded environments of the cell</b>
82	<b>8.3. Need to develop parameters for post-translational modifications</b>
83	<b>9. Conclusions and perspectives</b>
84	
85	
86	
87	
88	
89	
90	
91	
92	
93	
94	
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## 99 1. Introduction

100

101 Modern molecular biology experiments involve *in vitro* studies to analyse the biomolecules in  
102 isolation in a test tube, the site-directed mutagenesis and transport or interactions of small  
103 molecules with the proteins. These investigations record the changes occurring at macro time  
104 scale but the microscopic details that lead to these macro level changes still remain a mystery.  
105 The Molecular dynamics (MD) of these biosystems in a simulation box provides a virtual  
106 microenvironment for analysing them in minute details. MD simulations with atomistic, physical  
107 parameters-based force fields involve principles of thermodynamics that may offer the potentials  
108 to gain new insights into the functional mechanisms operating at the level of single biomolecule  
109 to the biosystems (at the organelle levels) that are time and cost effective as compared to the  
110 conventional molecular gene expression, enzyme assays, protein isolation, characterization and  
111 crystal structure determination (Fig. 1) [1]. Several sampling methods are used for performing  
112 MD experiments on the biomolecules. These methods include adaptive umbrella sampling,  
113 integrated tempering sampling, metadynamics, accelerated MD and replica exchange, where MD  
114 has provided more robust conformational analysis possible for any of the biomolecules [2]. MD  
115 is based on the movement of biomolecules in the solvent box which follow Newtonian laws of  
116 motion. The spatial distribution of atomic properties of biomolecules is defined by force fields  
117 (FF) and its accuracy is essential for the success of an MD simulation experiment. All the atoms  
118 of a biomolecule (protein, lipid membrane, bound metal ions, associated water etc.) analyzed in  
119 an MD experiment are represented in the form of actual coordinates which generates new set of  
120 coordinates for the particles in terms of the energy of the system [3]. The parameters for force  
121 fields are developed by *ab-initio* calculations [4, 5]. The parameters of the FFs are adjusted in the  
122 light of experimental data obtained from neutron, X-ray and electron diffraction and scattering,  
123 NMR, infrared, Raman and neutron spectroscopy, etc. are used to generate the parameters for the  
124 atoms of the proteins. These parameters may also be obtained from *ab initio* or semi-empirical  
125 approaches of quantum mechanical calculations [6]. The atoms of biomolecules or a biosystem  
126 may be represented as simple elastic (harmonic) forces and FF replace these potentials of the  
127 system by a simplified model that is valid in the region being simulated [7,8]. However a typical  
128 expression for a FF may look like the following [9]:

129

$$E_{\text{total}} = \sum_{\text{total}} K_r (r - r_{\text{eq}})^2 + \sum_{\text{angles}} K_\theta (\theta - \theta_{\text{eq}})^2 + \sum_{\text{dihedrals}} \frac{V_n}{2} [1 + \cos(n\phi - \gamma)] \\ + \sum_{i < j} \left[ \frac{A_{ij}}{R_{ij}^{12}} - \frac{B_{ij}}{R_{ij}^6} + \frac{q_i q_j}{\epsilon R_{ij}} \right].$$

130  
131 Different mathematical symbols denote the various potential energy functions ( $E_{\text{total}}$ ) which  
132 include equilibrium terms and force constants. Equilibrium term  $r-r_{\text{eq}}$  describes the energies of  
133 deformation of the bond length and  $\theta-\theta_{\text{eq}}$  represents the bond angle. The third term represents  
134 rotations around the chemical bond that are characterized by periodic energy terms (with  $n$   
135 dihedral multiplicity, designated as  $n$  and steric hindrance defined by  $V_n$ ),  $\Phi$  is the dihedral angle  
136 and  $\gamma$  is the phase during rotation around the dihedral angle. The fourth term describes the van  
137 der Waals interatomic forces in the form of the Lennard–Jones (L-J) 12-6 potential, and the last  
138 term represents the Coulombic electrostatic potential. Some effects due to specific charge  
139 environments can be included using properly adjusted partial charges  $q_i$  as well as the van der  
140 Waals parameters ( $A_{ij}$  and  $B_{ij}$ ) and force constants i.e.  $K_r$ : bonds;  $K_\theta$ : angle (Fig. 2) [10]. Here,  
141 the first four terms are referred to the local contributions (intramolecular) of the total energy  
142 within the molecules (bond stretching, angle bending and dihedral & improper torsions), and the  
143 last two terms describe the repulsive and Van der Waals interactions (in this case using a 12-6 L-  
144 J potential) and the Coulombic forces respectively (Fig. 2). All these parameters for FF are  
145 defined earlier and the parameters for the side chains of the biomolecules are determined at the  
146 end of the simulation experiment [11]. It is observed that the current FFs overestimate the  $\alpha$ -  
147 helical fraction of the protein when compared to the experimental J-couplings [7]. Therefore, the  
148 accurate balance between the secondary structure propensities of protein is a central priority for  
149 the FF development for the protein because it may be considered as the readout for the protein  
150 folding, aggregation, association, and conformational changes which may be studied by MD.  
151 AMOEBA (Atomic Multipole Optimized Energetics for Biomolecular Simulation) FF for  
152 proteins has been developed to study permanent electrostatic multipole moments through the  
153 quadrupole at each atom which explicitly treats polarization effects in various chemical and  
154 physical environments [12]. However, there is need to further improve AMOEBA FF should also  
155 consider short-range electrostatic and Vander Waals components and their coupling to valence  
156 interactions that may improve this polarizable FF further. The improved FFs may use multicentre  
157 atomic charges and addition of the polarizability for backbone hydrogen bonds. Therefore, it  
158 would be necessary to change the balance of interactions occurring in the water model, which  
159 would be affected by any further changes. To keep it constant, it would be effective to introduce  
160 parallel terms for protein-solvent and solvent-solvent interactions [13]. Binding of two  
161 compounds/biomolecules follows the fundamental principles of physics i.e. molecular diffusion  
162 and intermolecular forces. However, the processes that occur inside the cells consist of many  
163 components of the cell in a single chemical reaction (isomerisation, ligation, hydrolysis,  
164 synthesis, transport etc.) [10,14]. A simple chemical reaction of one enzyme and substrate that  
165 occurs inside the cells becomes further complicated by the internal dynamics of the enzyme

166 (solvent accessible area, flexible and stable regions of the enzyme and the location of the binding  
167 site on the enzyme). The changes occurring in the dielectric properties of the proteins and at  
168 interfaces are poorly understood and the complex cellular environment possesses large changes  
169 due to the very large solute concentrations inside the cells [1,15]. It is important to note that the  
170 aim of this work is not to review all the available literature, in the traditional way. The motto of  
171 this article is rather to explore the interdisciplinary areas of computational biology and  
172 bioinformatics and their applications in other allied fields namely, biophysics and biochemistry  
173 from the view of a researcher of basic biology. This work may also provide a better  
174 understanding of the concepts of the MD simulations of the biomolecules protein, nucleic acid  
175 (DNA/RNA) [16] membrane lipids [17], carbohydrates, metal ions, nanoparticles etc. at the  
176 atomic levels and to further understand the possible mechanisms operating inside the cells  
177 [14,18-22] (Fig. 1). However, in contrast to the wealth of structural data (>1, 23,000 structures in  
178 the Protein Data Bank as retrieved on 10/16/2016), accurate information on different  
179 biomolecules related to their behaviour and dynamics is much more sparse. Therefore, the  
180 development and validation of simplified MD protocols and FFs still remains a challenging task.  
181

## 182 **2. Different force fields for studying Biomolecules**

183  
184 There are different FFs available to carry out MD studies. These include AMBER, CHARMM,  
185 GROMOS, and OPLS Force Fields [1,10,20,23]. There are also ongoing constant efforts to  
186 improve the existing FFs. These include the optimization of parameters for the biomolecules in  
187 terms of developing more robust mathematical expressions especially for the torsion angles  
188 between the two sets of three atoms having two atoms in common among them. Moreover, the  
189 integration of multiple time-steps based on dual Hamiltonian propagators will help to observe  
190 quantum mechanics/molecular mechanics (QM/MM) at a faster rate [24]. The AMBER package  
191 provides parameters for more composite molecules and fragments not currently available in the  
192 other FF libraries. Recently L-J parameters as well as the bond and angle parameters were  
193 refined for the General Amber Force Field. General Automated Atomic Model Parameterization  
194 was used to reparametrize the electrostatic and dihedral terms [25]. Yoo and Aksimentiev (2016)  
195 refined the L-J parameters describing non-bonded interactions between specific nitrogen-oxygen  
196 atomic pairs using CHARMM and AMBER FFs for amine-carboxylate, amine-phosphate, and  
197 aliphatic carbon-carbon interactions that provide remarkable equivalent results of MD  
198 simulations of proteins, nucleic acids and lipids with experiments [26]. The AMBER package  
199 released in 2012 also supports a wide range of FFs for the analysis of different molecules  
200 including the non-native variants of amino acids and the nucleic acids. Further it also included a

201 new fixed-charge with enhanced support for polarizable potentials. AMBER FF (ff99SB protein  
202 FF) was mainly meant for protein secondary structure balance and dynamics. The amino acid  
203 residues namely, isoleucine, leucine, aspartate and asparagine that exhibited particularly large  
204 deviations from the experimental distributions, indicated that the side chains of these residues  
205 were not modeled properly. Therefore, the improved version of AMBER ff99SB-ILDN has been  
206 proven to be better than that of the earlier ff99SB version [27,28]. However, new protein FF i.e.  
207 RSFF2 (residue-specific force field) based on protein coil library of 20 native amino acids was  
208 developed by modification of the AMBER ff99SB FF that has provided a better secondary  
209 structure balance for  $\alpha$ -helical peptides,  $\beta$ -hairpins and Trp-cage, thereby, increasing the stability  
210 of  $\alpha$ - helix and  $\beta$ -sheet simultaneously [2]. Recently Zgarbová et al. (2015) have refined the  
211 parameters for the sugar-phosphate backbone torsion  $\beta$  for AMBER FFs to improve the  
212 description of Z- and B-DNA (Fig. 2) [29]. The MD with explicit solvent was carried out  
213 including the refinement of the  $\epsilon/\zeta$  and glycosidic torsions. Further, sugar-phosphate backbone  
214 torsion angle  $\beta$  ( $\beta_{\text{OLI}}$ ) was also refined so that it considered conformation-dependent solvation  
215 effects. This improved  $\beta_{\text{OLI}}$  was significantly able to model and describe the structure of the Z-  
216 DNA and its backbone as well as antiparallel guanine quadruplex (GDNA). It was slightly  
217 capable of describing the sub-states and also significantly described the sugar-phosphate  
218 backbone equilibria of B-DNA [29]. The accuracy of three current FFs in modeling the  
219 disordered peptides was assessed by combining enhanced-sampling MD simulations with NMR  
220 data. AMBER03w showed on an average more accuracy in describing the disordered systems.  
221 Moreover, it can be improved by incorporating the side-chains ILDN corrections that were  
222 developed for the AMBER99SB force field [28]. Best et al. (2012) have optimized the backbone  
223 correction energy map (CMAP) correction parameters with new side-chain parameters from  
224 complex quantum mechanics (QM) calculations for CHARMM36 [11]. The earlier available  
225 OPLS-AA/L FF reproduced potential mean forces well for some hydrophobic residues (such as  
226 Leu, Phe, Tyr, Trp, Val, Ile). However, it could not demonstrate the rotamer distributions  
227 (relative free energies of the three minima) well for Glu, His, Cys, Ser, Thr, and Asp. Therefore,  
228 the parameters shared between different  $\chi$  torsions and different amino acid types (Fig. 1) i.e.  $\chi_3$   
229 and parameters of non-hydrogen atoms were adjusted for better description of the rotational  
230 barriers in the improved version of the OPLS-AA/L FF i.e. residue specific force field1 (RSFF1).  
231 Though, RSFF1 provided a considerable balance between  $\alpha$ -helical and  $\beta$  -sheet structures, it



232 also described well the transferability among different sequences [30]. The measurement of  
233 inter-structure distance provides the way for calculation of the flexibility and disordered regions  
234 of protein. This utility is present in GROMACS and provides a basis for multidimensional  
235 scaling and conformational clustering [31]. Moreover, there is further need of improvements in  
236 the current FFs, since there are several limitations in them which are discussed in the following  
237 sections.

238

### 239 **3. Points to be taken care of before embarking on a molecular dynamics experiment**

240

241 Before starting the MD studies on any of the biomolecules, we should be well acquainted with  
242 the literature available about the molecule of interest starting from the very beginning. Different  
243 factors should be considered such as availability of its tertiary structure in the PDB (Protein data  
244 bank), its existing monomeric and oligomeric states, biological unit and different conformational  
245 states should be taken into consideration. The conformational changes occurring in the binding  
246 site of the protein are stimulated by the action of additional inducers like co-factors [metal ions  
247 ( $\text{Na}^+$ ,  $\text{K}^+$ ,  $\text{Mg}^{2+}$  etc.), ATP, NADP etc]. If the protein is a crystal structure, then the structure  
248 which was refined at resolution better than 2.5 Å should be used [32] and all type of protein  
249 structures should be validated by Ramachandran plot based methods such as PROCHECK,  
250 MOLPROBITY etc. [33-35]. PROCHECK and MOLPROBITY use the distributions of  $\chi$  side-  
251 chain torsion angles to validate the protein structure. The overall “clashscore,” which represents  
252 the number of bad all-atom clashes per thousand atoms (including hydrogen) should be less than  
253 five. The outlier amino acid residues should not be more than 5% in the Ramachandran plot [36].  
254 The di-sulphide bonds in the proteins should be analyzed as these provide stability to the protein  
255 tertiary structures and play major role in the folding and unfolding of the proteins. The header of  
256 a PDB file contains information about the coordinates of atoms and quality of protein structure  
257 [resolution, completeness,  $I/\sigma$  (or signal-to-noise ratio) and the highest resolution shell] as well  
258 as indicators of the quality of the resulting structure, such as refinement parameters i.e. R-factor  
259 and R free obtained from refinement programs like Refmac or Phenix [37]. If the protein  
260 structure available in PDB is determined in solution using NMR spectroscopy, then, the data  
261 available in PDB about the *B*-factors of NMR models comprising of the root mean square  
262 deviations (RMSD) from average ensemble coordinate positions should be checked. The

263 biomolecular structures which show RMSD below 1.0 Å in the side-chain values, main-chain  
264 deviations should not be greater than 0.4 Å. The deviations in bond length and bond angle should  
265 be of the order of 0.018 Å and 1.1 Å respectively [38]. The chemical shifts, residual dipolar  
266 couplings and relaxation rates provided in NMR data indicate the quality of the protein structure  
267 [39,40]. Ideally, the non-protein atoms (HETATM, SOL and ions) should be removed from the  
268 PDB file. It should also be noted that protein structure is in its native form or a mutated version.  
269 If the structure represents a mutated version, then it can be modified to its native form using  
270 PyMOL/Coot [41,42]. The protein structure contains the whole biomolecule or if some parts of  
271 the biomolecule are missing, it should be checked and the missing parts should be repaired.  
272 Since the interactions of the biomolecules are described as bonded and non-bonded interactions  
273 and are defined in a set of parameters, therefore, it is necessary that all the atoms ( $\alpha$ -carbon,  
274 polar donor, polar acceptor, apolar, and ionic) of the biomolecule should be explained by the  
275 empirical parameters of the FFs and there should not be any unusual behaviour of the atoms or  
276 steric clashes. It can be examined by visualizing the protein structure with graphical visualizers  
277 like Coot or PyMol that can check the amino acid residues which are lying outside the allowed  
278 regions of Ramachandran plot, and can bring them to the allowed regions by refining the  $\phi$  and  $\psi$   
279 angles of the outlying amino acid residues [41,42]. The overall charge of the system should be  
280 neutralized before the energy minimization. The water model in which protein has to be  
281 immersed should be selected cautiously as the loop and fold regions behave differently and the  
282 folding/unfolding processes affect both the equilibrium and dynamic properties of the system  
283 [43]. It can be concluded on the basis of the comparative structural behaviour of protein in  
284 different water models [44]. Among these, A99 and C22 of TIP4P/2005 were recently ranked as  
285 the most accurate force-fields for observing beyond microsecond time scale analysis [43]. For  
286 carrying out MD experiments, energy minimization, equilibration and production runs are  
287 usually carried out in steps as depicted in the flowchart (Fig. 3). GROMACS is one of the freely  
288 available **suites** which can be used and MD simulation is an effective tool which may provide  
289 crucial information about protein-ligand, protein-protein, post-translational modifications of  
290 proteins, protein-DNA/RNA and protein-lipid (embedded in membrane) interactions [45].  
291 GROMACS provides different utilities like `g_energy`, `g_rms`, `g_rmsf`, `g_hbond`, `g_dist`, `g_sas`,  
292 `g_density`, `g_isd` etc to obtain this information. These are used to analyze the trajectory files  
293 obtained at the end of MD sessions and crucial conclusions could be drawn. The trajectory file

294 obtained at the end of MD should be analyzed by any molecular viewer such as Visual Molecular  
295 Dynamics (VMD) and if the native structure of the biomolecule distort/break during the  
296 simulations, it indicates the structure of biomolecule is not stable [46]. In such a situation, there  
297 is need to check every step and the parameters applied on the biomolecule and similar studies  
298 conducted on same biomolecules should be studied before carrying out the simulations again. All  
299 points described above should be kept in mind before proceeding for an MD experiment. Further,  
300 the duration of MD will depend on the type of biological system under the study. In case of  
301 molecular dynamics of protein, it depends on the length of amino acid sequence. Other factors,  
302 which may be taken into consideration for deciding the simulation time, may be folding-  
303 unfolding status and effect of mutation on the protein structure etc. For instance, protein-small  
304 molecule interactions of an average size of globular receptor can be analyzed well in 100 ns  
305 duration of MD session [47,48]. For another case, a membrane efflux protein was analyzed, the  
306 transport of ligand across the membrane through transporter protein is relatively longer  
307 phenomenon to capture, in such situation, the analysis which could be computationally less  
308 expensive and serve the purpose have been used by the researchers in smaller time scale to show  
309 only the direction of movement of the ligand towards the extracellular space after it was picked  
310 by the binding site at the intracellular side [49]. For MD studies of protein-protein interactions,  
311 the interacting sites at the interface of the protein-protein complex should be selected on the  
312 basis of already published and experimentally validated data. The potentially interacting proteins  
313 may be analyzed for the predicted disordered regions. It has been well established that,  
314 disordered regions, loops and turns are prone areas of molecular interactions between any two  
315 interacting partners [50]. There are online servers like Frustratometer, which compute the  
316 possible sites for protein-protein interactions as readout of frustrated structural parts of the  
317 proteins [51,52]. Some of the specific applications have been discussed in the following sections  
318 in detail.

319

#### 320 **4. Molecular dynamics of protein-ligand systems**

321

##### 322 **4.1. Apo protein and protein-small molecule complex MD analysis**

323 MD of protein in apo form and protein-ligand complex may be carried out at the atomic levels

324 e.g. in the case of apo protein, the long MD analysis provided atomistic detail about different

325 folding states of the protein. Further, the enzyme catalyzed reactions where proteins interact with  
326 small molecules which cannot be observed by the conventional *in vitro* approaches like X-ray  
327 crystallography was demonstrated by MD analysis [55, 56]. The present FFs are well developed  
328 for studying the various functions of biomolecules (catalysis, ligand-binding locations, transport,  
329 motility, signal transduction, allosteric regulation, and molecular recognition) (Fig. 4). MD  
330 studies are widely used for the prediction of stability of the potential mutants of a protein  
331 molecule with higher catalytic efficiency or lower binding affinity for the toxins/molecules that  
332 may interfere with their normal function [35, 57, 58]. The modifications in the transporter  
333 protein's conformation by the normal motion patterns caused by the surrounding solvent or by  
334 the binding of a ligand molecule may be correlated with the efflux mechanism. These changes in  
335 the structural conformation of the transporter caused by the ligand binding are well studied by  
336 MD approach [14, 49]. A comparative MD simulation study on the NS3 protease of Hepatitis C  
337 virus has illustrated that the catalytic triad presents a dynamic behaviour which triggers a  
338 conformational instability in the protein that finally leads to abortion of the drug binding against  
339 the virus [59]. MD analysis of mutated HIV-1 IN protein (trans-activating regulatory protein) of  
340 the virus complexed with L-731,988 ( $\beta$ -diketo acids) could reveal the binding modes of this  
341 ligand and also helped in understanding the mechanism of its inhibition [60]. Natarajan and  
342 Senapati (2012) demonstrated the interactions and binding of the drugs (taxol and epothilone)  
343 with the mutants of tubulin dimer protein using MD simulations [61]. It was observed that the  
344 mutations primarily modified the M loop conformation of the protein primarily, enlarged the  
345 pocket volume and affected the distal sites of tubulin involved in the microtubule building  
346 processes [61]. The enzyme inhibitor process was reconstructed by MD simulations of free  
347 ligand binding that involves the placement of a ligand at certain distance from the protein of  
348 interest. The ligand binds first by diffusion, followed by specific interactions with protein. It  
349 binds to one or more sites in the protein. Benzamidine binding to trypsin showed various states  
350 during MD simulations [62]. Therefore, MD based studies have accelerated the research progress  
351 in the field of biomedical research by providing easy, fast and quantitative method for analysing  
352 the protein-drug interactions. The MD has helped in deciphering the behaviour of the drug in the  
353 binding pocket of the target protein, its stability and specificity. Hong et al (2016) have  
354 employed the integrated approach involving neutron spin echo (NSE) spectroscopy and MD to  
355 elucidate the collective movement of the domains which provide access to the catalytic site [64].

356 Therefore, the structural studies of protein and protein-small ligand molecule which will involve  
357 integrated strategies of *in vitro* and *in silico* may lead to better results [53, 54].

358

#### 359 **4.2. Protein-protein interaction specificity and binding**

360 The interactions between proteins occur mainly by diffusion, non-covalent forces and the  
361 conformational flexibility present in the proteins. These are intertwined in such a way that they  
362 may be elucidated by MD simulations at atomistic levels. The non-covalent interactions  
363 occurring between biomolecules are the foundation of cellular biochemistry. The binding of two  
364 protein molecules is convoluted by complex internal dynamics with constantly changing  
365 dielectric properties inside and around the protein molecules. It is necessary to quantitatively  
366 access the binding region of the protein receptor and ligand so that the particular region of the  
367 protein involved in the interactions should be considered for MD as described in this article in  
368 earlier section. Moreover, the redistribution of solvation energy among the proteins and at the  
369 interfaces is poorly understood but they may be involved in protein-protein interactions or in  
370 crowded environment of the cells that represent varied solvent accessible surface area and  
371 dynamic solute concentrations inside the cells. Many of the causative forces act simultaneously  
372 in protein-protein interactions, making it even more complex phenomenon to analyze when the  
373 new incoming molecules appear in the analysis frame, rotational reorientation and  
374 reorganization, and steps of conformational selection and/or either induced fit or lock-key model  
375 hypotheses of the potential complexes may be operational. The protein-protein interactions of the  
376 biotin–streptavidin complex have been studied *in vitro*, however the atomistic details about the  
377 binding, optical transitions, harmonic vibrations and absorption spectra was obtained by  
378 molecular mechanical methods [65]. MD simulations can provide details upto atomic levels with  
379 high accuracy in nanoseconds to microseconds time scale in protein-protein interactions.  
380 However, recent studies have clearly demonstrated the importance of incorporating electrostatics  
381 with multiple charges, variable dielectrics and polarizability in atomistic force fields, especially  
382 for the better reproduction of hydrogen bonds, water defects/effects, aromatic chain stacking, pi-  
383 pi interactions, ion-aromatic ring interactions and other non-bonded interactions. It is commonly  
384 observed that in MD studies with the recent versions of the AMBER, CHARMM and OPLS FFs,  
385 the unfolded proteins tend to become much compact in simulations as compared to the  
386 experimental data [66]. While it is reported by MD analysis that the protein-protein interaction

387 specificity and binding advances by superposition of conformational selection and induced fit  
388 starting at least at surface-to-surface distances of 10 to 20 Å. Abriata and Peraro (2015)  
389 demonstrated that the conformational selection slowly occurs first and shifts to induced fit as the  
390 proteins move in close proximity[1]. Then, antiparallel orientation is adopted by dipolar  
391 moments and finally intermolecular contacts are established [1]. Antigen processing and antigen-  
392 antibody interactions are most important processes in the adaptive human immune response.  
393 When signalling processes occurring at the cellular levels were studied by MD approaches, they  
394 gave an insight into the diverse binding behaviour of the major histocompatibility complex  
395 (MHC) with different peptides. Different MHCs bind to same peptide which is aided by the  
396 conformational changes caused by the binding of different T cell receptors (TCRs). The MD  
397 studies can be used to explore the interactions of TCR-peptide-MHC ternary complex, the  
398 structural behavior of MHCs without ligand and simulations of the MHC containing trans-  
399 membrane regions. Further steered MD of TCR bound to pMHC and structural refinement of the  
400 homology modeled MHC proteins by MD may help to decipher their diverse binding modes. The  
401 conserved binding pattern of TCR/pMHC could be due to co-evolution of key amino acids  
402 involved in the interactions which can be studied in detail using MD [21]. A chimeric subunit  
403 model vaccine was designed using structure based rational approach. Thermodynamic stability of  
404 the vaccine was enhanced by disulphide engineering and its conformational accuracy and  
405 stability was assessed using MD simulations [21]. The integrated strategy comprised of  
406 equilibrium NMR analysis of thermal unfolding and long MD simulations deciphered the folding  
407 of W protein of bacteriophage lambda (gpW). It showed two-state-like fast folding dynamics and  
408 cooperative equilibrium unfolding behavior. It was observed that a complex pattern of structural  
409 changes occur at the atomic level. These changes result in detailed network of residue-residue  
410 couplings that lead to cooperative folding emergence [67]. However, the integrated approach  
411 which involved Terahertz spectroscopy and MD study of aqueous peptides showed the depth of  
412 hydration water around proteins and demonstrated the protein-protein interactions [63].  
413 Therefore, MD approaches integrated with crystal structural studies may be used to decipher the  
414 possible mechanisms involved in the protein-protein interactions, specificity generation and the  
415 conformational changes occurring at the interface of the two proteins. It may lead to the  
416 stimulation, propagation and completion of the cellular processes.

417

### 418 **4.3. Behaviour of nucleic acid-binding proteins**

419 DNA-binding proteins take part in various cellular and molecular processes ranging from DNA  
420 polymerization to DNA repair and proof-reading, transcription and DNA super-coiling  
421 regulation. The extra-ordinary specificity and efficiency of DNA-binding proteins present a  
422 major theoretical riddle due to their enormous variations in the length and amino acid sequence  
423 interacting the DNA molecules. This fact is further supported by studies on protein–DNA  
424 complex structures. Those have explained about the molecular basis of specificity in individual  
425 or highly related complexes. However, there are no specific signature sequences or motifs have  
426 been found for a universal or generic recognition code which may adequately explain the  
427 observations about the specificity of all DNA binding proteins [68]. The sliding movements of  
428 DNA-binding protein and an RNA-binding protein (Barnase) along the B-DNA were studied  
429 earlier using MD simulations [69]. It was recorded for the protein-DNA interactions, there is lack  
430 of bonded interactions and only electrostatic and H-bonding interactions are prevalent. It was  
431 also observed that the protein molecule’s movement was bidirectional in agreement with NMR  
432 data [69]. The contribution of dynamic behaviour of a single amino acid chain for the  
433 identification of one or more base pairs of telomere repeat binding factors (TRF1 and TRF2) that  
434 may help in resolving the conflicting experimental data was also explained by MD studies [22].  
435 It was demonstrated by MD that two arginine side chains oscillated and induced direct  
436 interactions between DNA bases, DNA back bone and the transcription factor SKN-1.  
437 Disappearance of the protein–DNA salt bridges was observed with sub-nanosecond timescales of  
438 MD [69]. The conformational transitions in cyclic AMP (cAMP) receptor protein from  
439 *Mycobacterium tuberculosis* while binding to DNA was also analyzed by MD and normal mode  
440 analysis. It demonstrated that different orientations between the cAMP-binding and DNA-  
441 binding domains arise due to significant movement in the residues just before the C-helix and a  
442 larger movement at the end of C-helix. MD enabled us to propose a universal mechanism of  
443 allostery-mediated DNA recognition of the CRP [70]. MD analysis of DNA structure in free and  
444 bound to Cren7 (chromatin protein) protein at different temperatures was carried out. It was  
445 observed that dsDNA was stable in certain range of temperatures. It was also observed that some  
446 residues of protein have significantly formed H-bond and hydrophobic contacts with the DNA  
447 which contributed to the stability of the molecule. It was also reported that DNA got transitioned  
448 from B-like to A-like form with an increase in temperature [71]. Similar results were also

449 reported for the stabilization of DNA by Sso7d (hyperthermophilic proteins) in *Sulfolobus*  
450 *solfatarius* and further, it was found that only eight amino acid residues (Lys7, Tyr8, Lys9,  
451 Lys22, Trp24, Val26, Met29 and Arg43) were significant for the formation of the binary  
452 complex with the DNA [72] (Chen et al., 2012b). The intercalation of small drugs into DNA was  
453 also elucidated using MD and a mechanism of “detrapping” paradigm was explained in which  
454 minor grooves of DNA are involved in binding and also acting as inhibitors for the functional  
455 DNA-protein complexes. While intercalator molecule binds to DNA, there is a small energy  
456 barrier to overcome which favors unwinding of the DNA [73]. Metadynamics studies have  
457 shown that drug-DNA interactions are complex processes that elaborate the navigation of many  
458 basins (target) by the drug to non-specific contacts before reaching the significant pocket/groove  
459 exhibiting high binding affinities [74]. The alteration in the nucleic acid structure can be mainly  
460 of three types in DNA-protein complexes namely, DNA structure is not altered, the duplex  
461 integrity is consistent but DNA structure is disrupted and finally overall duplex organization is  
462 lost. The DNA Y-family polymerase IV (DPO4) protein of *Sulfolobus solfataricus* undergoes  
463 conformational change from an “open to close” state while interacting with DNA. MD analysis  
464 showed that DPO4 transitioned between multiple states while interacting with DNA and  
465 conformation distribution varied at different binding stages that regulated the DNA recognition  
466 [75]. MD was performed on the consensus sequence of DNA present in the (tumor suppressor)  
467 p53-specific binding site which elucidated that binding of p53 to DNA caused a local DNA  
468 conformational change and apparent DNA bending [76]. The DNA-protein interactions may  
469 occur due to the formation of hydrogen-bonds specifically between the protein and nucleotide  
470 bases, grooves mediated electrostatic forces (depending on the geometry and sequence of  
471 DNA/protein) and conformational changes due to sequence specificity [77-80]. Trichothecene  
472 molecules were reported to inhibit protein synthesis and MD analysis showed that trichodermin  
473 interacted with the nucleotide residues of 25S rRNA that constitute the peptidyltransferase centre  
474 (PTC). However, the trichodermin resistance protein (60S ribosomal protein L3) was reported to  
475 overcome the inhibitory effects of trichothecene molecules. Further, MD of trichodermin  
476 resistance protein and 25S rRNA consisting of PTC showed that the W-finger region of the  
477 protein moved towards 25S rRNA and blocked the binding pocket of the trichodermin [81]. The  
478 large conformational change caused by binding of U1 (nuclear ribonucleoprotein) protein with  
479 RNA was investigated using MD. It was observed that there are modifications in hairpin and



480 internal loop of the protein upon binding to the RNA [82]. Therefore, MD elucidated the  
481 mechanisms and conformational changes induced by the DNA-binding proteins at the particular  
482 binding site or the minor/major groove of DNA/RNA involved in the cellular functions.

483

#### 484 **4.4. Post-translational modifications of proteins**

485 The post-translational modifications (PTM) (addition or removal of chemical groups like  
486 phosphates) occur at the side chains of Ser, Thr, and Tyr in all organisms. There are several  
487 enzymatic and non-enzymatic PTMs. More than 250 distinct PTM reactions or 110 non-  
488 redundant PTM amino acids and protein termini have been reported [83]. The charge carried by  
489 phosphate is predominantly -2 at relevant pH of the cell. This results in large changes in  
490 electrostatic interactions that direct the energy landscapes influencing the folding of the protein  
491 to protein interactions and interactions with ligands that may also modulate its catalytic  
492 activities[84]. Groban et al. (2006) described the changes in the free energy landscape caused by  
493 phosphorylation of CDK2 which leads to conformational change in the activation loop of the  
494 protein using MD analysis [85]. The effects of phosphorylation could be observed on the  
495 immediate adjoining local structures of the protein. The phosphorylation like PTM may lead to  
496 repositioning of the interaction network in adjoining regions of the protein. There are some  
497 reports which showed that the conformational differences in the active and inactive forms of  
498 kinase were observed by MD simulations performed on Hck, an Src family kinases [86]. It is  
499 documented that few proteins contain hyperphosphorylated peptide regions i.e. various residues  
500 phosphorylated in close proximity e.g. hyperphosphorylation has been reported in the  
501 microtubule-associated protein tau which is commonly found in Alzheimer's disease [87]. The  
502 effect of hyperphosphorylation on the structural behavior of proteins is not well elucidated.  
503 Hamelberg et al. (2007) demonstrated that the hyperphosphorylation induced unfolding of the  $\alpha$ -  
504 helix that leads to the formation of a compact 'arginine-claw' structure [88]. A single phosphate  
505 coordinates with five to six arginine residues while the other phosphates are solvent-exposed, a  
506 characteristic of arginine-claw. The folding mechanism of Kinase-inducible domain (KID)  
507 (transcriptional activator) was elucidated by the MD approach which showed that KID  
508 transforms a disordered protein to well folded protein by the changes induced by  
509 phosphorylation after post-translational modification [89]. MD analysis of ubiquitin protein  
510 showed long-range correlated motions which linked distal residues that resulted in the binding of

511 the interface of a protein. Further, this insinuated a mechanism which may lead to  
512 conformational selection, molecular recognition and biological function of ubiquitin [90]. It is  
513 demonstrated that myristoylation resulted from the interaction of myristate to N-terminal glycine  
514 residue [91]. It is shown that many proteins of eukaryotes are acetylated at the N-terminal  
515 residues namely, Ala, Ser, Val, Thr and Cys. Histone deacetylases complexed with its inhibitor  
516 showed the interactions active site residues with inhibitors using MD analysis which regulate the  
517 gene transcription [92]. The parameters were developed for PTMs which have been able to  
518 reproduce experimental hydration free energies (HFEs), a measure of hydrophobicity, which is  
519 one of the most important properties of the amino acids that drives protein folding, ligand  
520 binding or protein-lipid interactions. The parameters were generated for GROMOS FF (45a3 and  
521 54a7) for over 250 diverse enzymatic and non-enzymatic PTMs because these parameters were  
522 fitted to reproduce experimental HFEs. This tool is available online for studying PTM with MD  
523 (<http://vienna-ptm.univie.ac.at/>) [83,93]. Although not enough experimental studies are available  
524 on following seven PTMs namely, carboxyllysine, homocitrulline, S-carbamoyl-cysteine,  
525 citrulline, 2-oxo-histidine, S-nitrosocysteine and pyruvic acid [83], however, other post  
526 translational modifications such as methylation, sulfonation, myristoylation, acetylation,  
527 ubiquitination, glycosylation and methylation of amino acids are yet to be explored by MD  
528 analysis [94]. Post-translational modifications have important roles to play in controlling  
529 numerous eukaryotic cellular functions such as regulation of gene expression, chromosome  
530 structure and cell signalling. Only some of the post-translational modifications such as  
531 phosphorylation, acetylation, and ubiquitination have been studied in detail. While the  
532 mechanisms involved in the signal propagation by these modifications are unknown and hence,  
533 more attempts using combined techniques of *in vitro* MD are warranted to unravel the unknown  
534 mechanisms.

535

## 536 **5. MD analysis pertaining to lipid bilayer biomembranes**

537

### 538 **5.1. Physiological properties of plasma membrane analyzed by MD**

539 Lipids are very crucial components of the biological membranes and vary in their structure and  
540 function. Their biological roles range from signalling and energy storage to regulation of  
541 membrane protein functions. Although, experimental research is not yet fully efficient to unravel

542 the biological phenomenon of lipid functions to a great extent because of persisting difficulties in  
543 handling and isolation however, these limitations may be overcome with the application of  
544 molecular simulation techniques. The effect of presence of cholesterol (condensing effect) on  
545 lipid molecules in the form of change in the order parameter of lipid tails was recorded with the  
546 help of MD simulations [95]. MD simulations have also been able to explain lateral self-  
547 diffusion coefficient of lipid bilayer with changes in temperature and variation in the  
548 composition of the model membranes [96]. Nowadays, atomistic scale simulation, which is able  
549 to simulate bigger patches of lipid bilayer over a larger timescale, **is an advancement in the field**  
550 **of lipid MD analysis** [97].

551 These lipid membrane simulations showing lipid rafts and cholesterol interactions with the other  
552 lipid molecules have greatly changed the traditional outlook of lipid membrane. It has explored  
553 many more novel dimensions of lipid-lipid and lipid-protein interactions and their role in  
554 subsequent cellular signalling and trafficking across the cellular membrane. The major obstacles  
555 in the area of the lipid MD simulations are the accurate imitation of the asymmetric lipid bilayers  
556 with varied lipid composition. The coarse-grained models developed in the recent years have  
557 solved these problems to some extent [95,98]. Though understating the molecular driving forces  
558 for lipid domain formation still requires much more chemical details. However, many new  
559 developments in polarisable lipid FFs are in progress, which appear to be in harmony with the  
560 experimental data like area per lipid, electron density profile and deuterium order parameters of  
561 lipid membranes that will hopefully provide a more realistic view of the lipid molecules present  
562 in the biomembranes [99]. Furthermore, Lindorff-Larsen et al (2016) studied the self-assembly of  
563 the peptides with the phospholipid DMPC using combination of coarse-grained MD simulations,  
564 size-exclusion chromatography, circular dichroism spectroscopy, small-angle X-ray scattering,  
565 static light scattering and UV-Vis spectroscopy [56]. It revealed the molecular interactions  
566 occurring between the protein and lipids. Therefore it is needed that the biophysical studies  
567 should be carried out by the integration of experimental approach and computational techniques.  
568

## 569 **5.2. Conformational changes and functional mechanism of the integral membrane proteins**

570 The experimental techniques like the high-resolution X-ray structure studies cannot address the  
571 dynamic nature of conformational changes which may occur during the movement of membrane  
572 proteins. Therefore, *in silico* approaches have the potential to utilize the data from both the

573 structural and biochemical research and it may reveal the functional mechanisms involved  
574 [100,101]. MD studies of protein embedded into the lipid membranes should be carried out by  
575 considering the complexity of the system. There are well established FFs consisting of  
576 parameters for lipid membranes. The AMBER [20], CHARMM [102], GROMACS [45] and  
577 NAMD [103] suites are suitable to carry out MD simulation of membrane embedded proteins.  
578 MD of lipid membranes may unravel the behavior and dynamics of the membranes at the atomic  
579 and molecular levels, with different solvents, drugs, glycoproteins and transporter proteins that is  
580 not easily accessible in different biochemical experiments. It also helps in understanding the  
581 change in properties of lipid membranes due to interactions with diverse types of molecules [17].  
582 The first step in MD of protein embedded into the lipid membrane is the insertion of the protein  
583 into the lipid bilayer which mainly depends on the nature of the side chains of amino acid  
584 residues making the interface between the protein and the lipid bilayer. Further, the depth of  
585 insertion of the embedded protein in the lipid bilayer could be measured [104]. The solvent  
586 accessibility and other functional parameters are usually more sensitive to the distance from the  
587 polar surface of the membrane and therefore the distance between the phosphate groups in the  
588 lipid molecules and the centre of mass of the transmembrane helix in the inserted proteins are  
589 also quantified. The command `g_density` on GROMACS can be used to make a density profile  
590 for the head-groups and tail groups of phospholipids. It will be helpful to calculate the depth  
591 required to insert the protein in the lipid membrane. Recently, Javanainen (2014) developed a  
592 new protocol for embedding the proteins into the lipid bilayer by pushing them into the lipid  
593 bilayer membrane by the application of a high lateral pressure on the system from its sides [104].  
594 It does not distort the initial geometry of the lipid bilayer [105]. Till now, several integral  
595 membrane proteins have been analyzed using MD simulations for understanding their functional  
596 role. The arginine of the D/E/NRY motif in Rhodopsin family G protein-coupled receptors  
597 (GPCRs) is conserved and has its role in salt bridge formation and also in the maintenance of  
598 endoderm-1 (Tre1) structure as revealed by MD simulations [106]. The transport of various ions  
599 and other molecules across the lipid bilayer and the mechanism operating in ion transport have  
600 been described in detail at the atomic levels through MD simulations. These studies revealed that  
601 different mechanisms are involved in the transport of the sodium and chloride ions. The  
602 membrane adsorbed sodium ions diffuse across the interface while chloride ions move through  
603 the middle of the pore with the bulk water and a crossover from ion-specific permeation was

604 observed when the pore radius was increased [107]. MD can be used to decipher the solubility of  
605 the drugs and the pathway involved in their transport across the membrane. Free binding  
606 energies and the interactions of the drugs with the solute and membranes can be calculated and  
607 further, the intra- and intermolecular states of the drugs in the surrounding environment and the  
608 membrane can be characterized by MD analysis [108]. MD studies have revealed that the  
609 adaptor protein exhibited interdomain dynamics which is essential for the formation of the  
610 tripartite complex composed of an outer membrane porin protein, inner membrane pump and a  
611 periplasmic adaptor protein which is involved in the efflux of compounds from Gram-negative  
612 bacteria and induce drug resistance in the bacteria [49,109]. MD simulations can address  
613 physical principles underlying key membrane processes, for instance, the puzzle connected with  
614 channel high selectivity and present a pertinent question that how can the flexible and highly  
615 dynamic systems like proteins be functionally highly selective [110]? It is now documented that  
616 uptake and release of various substances at the cellular levels across the biological membranes in  
617 the organisms is affected by secondary-active transporters. These transporters have been reported  
618 to possess a fold having ‘inverted repeat’ motif consisting of five transmembrane helices  
619 arranged in two sets orientated oppositely with respect to the lipid bilayer. Based on MD analysis  
620 of the sodium-benzylhydantoin transporter protein, Mhp1, from *Microbacterium liquefaciens* a  
621 mechanism has been proposed for switching from the outward- to the inward- facing state which  
622 may lead to the inward movement of sodium-benzylhydantoin, that is primarily achieved by a  
623 stiff movement of some transmembrane helices compared to the rest of the protein [111]. The  
624 activity of agonistic molecules (carboxylic acids and non-carboxylic acids) on the G-protein  
625 coupled receptor (GPCR), GPCR40 was observed using MD. It demonstrated ligand binding  
626 mechanisms of different agonists and fluctuations in the position of these molecules docked  
627 inside the transmembrane domain of the protein embedded in the lipid membrane [112]. The  
628 binding behaviour of antagonistic molecule [ML10302 (4-amino-5-chloro-2-methoxy-benzoic  
629 acid 2-piperidin-1-yl-ethyl ester)] with the serotonin, h5-HT<sub>4</sub> receptor (GPCR) was investigated  
630 using MD. It revealed that the receptor formed key interactions with the ligand and water  
631 molecules were transported from the extracellular milieu towards the putative hydrophobic  
632 pocket [113]. There are several transporters involved in various cellular processes where the  
633 mechanism of transport remains still unknown. The synergistic hybrid applications of Double  
634 Electron Electron Resonance (DEER) data for molecular structure with molecular simulation

635 methods have been reported to be an efficient strategy to explore the conformational states in the  
636 membrane protein structure and function [114]. The restrained-ensemble and dummy spin-labels  
637 (DS) simulation methods were employed to obtain the desired refined structures using global  
638 ensemble-based restrained energies. The force spin-spin distance distribution and histograms  
639 were calculated from multiple-copy MD simulation to match those obtained from DEER. The  
640 DS simulation method is based specially on parameterized dummy spin-labels that mimic the  
641 dynamics of all-atom multiple time-steps, spin-label and provides distance distribution between a  
642 pair of spin-labels that can be directly compared to the distance distributions obtained from  
643 DEER to access the reliability or correctness of an already existing 3D model structure of the  
644 system [114]. The crystallography studies did not provide details about the actual open and  
645 closed conformational states of glutamine-binding protein (GlnBP) of *E. coli*. However, the  
646 combined strategy which involved NMR residual dipolar coupling analysis, MD simulations, and  
647 single-molecule fluorescence resonance energy transfer methods showed four metastable states  
648 co-existing for apo-GlnBP in solution [115]. Such hybrid and multi-technology assimilated  
649 strategies together with classical molecular mechanics may provide important insights into the  
650 conformational dynamics of ion channels and transporter proteins. These unravelled mechanisms  
651 can be elucidated by MD and may be helpful in understanding the cell signalling, transport of  
652 drugs across the biological membranes and may also provide the basis for combating pathogenic  
653 resistant strains of the microbes.

654

## 655 **6. Molecular dynamics of carbohydrates**

656 The structural organization of carbohydrates is different from polypeptides, proteins and nucleic  
657 acids, since they do not have well defined spectral regions that determine the secondary-  
658 structural motifs. Therefore, they do not have well-organized tertiary structures in solution.  
659 However, the conventional experimental methods such as X-ray crystallography have a long  
660 history of their application in research on carbohydrates. Yet, these methods provide a single  
661 three-dimensional model that cannot explain the dynamic properties and behaviour of  
662 carbohydrates [116-118]. Terahertz spectroscopy study showed that the long range hydration  
663 dynamics of anti-freeze glycoproteins lead to the anti-freezing activity in these protein [119].  
664 MD simulation has explored the multiple conformational states of oligosaccharides and  
665 glycoproteins. The FFs available for carbohydrates are GLYCAM06, CHARMM36 and

666 GROMOS 45A4. Various compounds can be studied by GLYCAM06, which is compatible with  
667 the AMBER MD package [120]. MD data were reported on carbohydrates and these include  
668 carbohydrate hydration, interaction of carbohydrates with ligands/other components,  
669 glycoproteins and other complexes of carbohydrates [121] (Fig. 4). Botulinum Neuro Toxin/B  
670 complexed with GD1A oligosaccharide and BoNT/B-GD1B complex was studied by MD which  
671 revealed different single binding mode of BoNT/B to GD1A i.e. single binding mode for GD1A  
672 and two binding modes for GD1B in the binding pocket of BoNT/B [121]. The dynamic  
673 behaviour of mannitol, sorbitol,  $\alpha$ -cyclodextrin, glucose, ribose and deoxyribose in aqueous  
674 solution has also been explored using MD [115,122-126]. The MD study of glycan chains and  
675 the glycoproteins were separately carried out which showed the conformational flexibility of  
676 glycan chains. It also provided the interactions of protein-carbohydrate and changes occurring  
677 after the intercalation of ligand at the atomic levels [126]. There is deficiency of well-resolved  
678 crystallographic coordinates of the carbohydrates. Therefore, it is not easy to measure the  
679 changes induced by the attachment of glycan chain and its conformation, structure and dynamics  
680 of the protein. Moreover, the multi-technology driven studies Raman optical activity spectra and  
681 MD and QM/MM approach carried out on D-glucuronic acid and N-acetyl-D-glucosamine that  
682 are constituents of hyaluronan [127]. These studies revealed the contribution of hydration  
683 interactions towards the conformational behaviour of these two carbohydrates [127]. MD  
684 simulation study of  $\beta$  (1 $\rightarrow$ 4)-linked D-aldohexopyranose disaccharides indicated that solvent-  
685 exposed intramolecular H-bonding in aqueous carbohydrates represents a minor (possibly  
686 adverse) conformational driving as well as steering force. The intramolecular H-bonding may be  
687 the cause of the dominance of secondary structure motif typical of cello-oligosaccharides [128].  
688 The disordered hydrogen bonding present in cellulose was studied using neutron crystallographic  
689 techniques and quantum mechanical calculations to elucidate the complex nature of hydrogen  
690 bonding in cellulose [129]. Further, studies involving such integrated approaches will provide  
691 understanding of the functioning of carbohydrates. All these processes can be studied in details  
692 at the atomic levels by integrated hybrid structural biology approaches using crystallography,  
693 SAXS, cryoelectron microscopy and MD of protein structure containing glycan chains.

694

## 695 **7. Molecular dynamics of DNA/RNA**

696 MD studies of nucleic acids are not very popular for various reasons. For instance, nucleic acids

697 are highly charged, non-globular and in many cases the crystal structures are not available (only  
698 1650 of DNA and 1190 of RNA PDB entries on 10/16/2016). The high charge densities on  
699 nucleic acids can be accurately simulated by polarizable FFs that are still in developmental phase  
700 [130,131]. Therefore, it is of importance to carefully develop parameters for the solvent and  
701 electrostatics [132]. Initial MD studies on nucleic acids revealed structural differences in DNA-  
702 DNA, RNA-RNA and DNA-RNA hybrid duplexes [133]. RNA duplex shows conformational  
703 switching in uracil-cytosine base-pairing mediated by water molecules [134]. In one  
704 conformation water acts both as H-bond donor and acceptor. The H-bond with uracil N3-H3  
705 donor site did not show much change during simulations while the H-bond between an OH group  
706 of water and cytosine N3 acceptor site of cytosine was found to rotate during the time of  
707 simulation. While other conformations were observed to be similar to water-mediated UC base-  
708 pair reported earlier in the crystal structure of the 23 S rRNA sarcin/ricin domain [134]. MD  
709 studies have revealed that RNA and DNA duplexes exhibit differential structural flexibilities for  
710 RNA. This varied structural behaviour of the RNA may be due to the local opening events  
711 occurring in the major groove of RNA while such incidence was not observed in the case of  
712 DNA. The solvent-base interactions, intra-strand hydrogen bonding among the RNA strands and  
713 improved rigid body motions of the RNA at the atomic levels is the cause of locally open  
714 conformations [76]. RNA hairpins that are an important class of RNA-RNA recognition motifs  
715 were found to have loop to loop interactions. MD of these RNA motifs revealed structural  
716 dynamics in the cation-binding pockets and major grooves that were reported to be involved in  
717 the dimerization of retroviral RNA genomes such as HIV-1 and Moloney murine leukemia virus  
718 [135]. Chen and García (2013) documented the *de novo* folding in case of three hyperstable RNA  
719 tetraloops [136]. The formation of non-canonical loop-stabilizing interactions among the RNA  
720 loops is the reason of this hyperstability. This **was reproduced and found** to be similar to  
721 experimental data during replica exchange MD simulations. This accuracy was obtained by  
722 reducing the L-J  $\sigma$ -parameters for base–base interactions, by adjusting the base–water  
723 interactions to achieve correct decomposition enthalpy and entropy for dinucleotides stacking  
724 and by the optimization of torsion angles of RNA [136]. MD analysis of DNA could be carried  
725 out for 0.1  $\mu$ s using the latest version of CHARMM [80]. The current MD simulations use  
726 explicit solvent and addition of counter ions (typically  $\text{Na}^+$  or  $\text{K}^+$ ) to achieve neutrality in the  
727 system. The umbrella sampling MD has revealed interesting mechanisms involved in bending



728 and base pair openings that may lead to local distortions involved in DNA repairing and  
729 epigenetic imprinting [132,137,138]. The folding/ unfolding of small DNA molecules can be  
730 studied within the  $\mu$ s time scale [80].

731

## 732 **8. Limitations of current force fields**

733

### 734 **8.1. Poor description of secondary structure of biomolecules by current force fields**

735 There are constant efforts to improve the present FFs. However, it was reported that  
736 CHARMM27 cannot fold all the WW domains of amyloid  $\beta$  protein due to over-stabilization of  
737  $\alpha$ -helical structures [139]. The parameters of backbone potentials were adjusted to reproduce the  
738 experimental J-couplings of Ala5 and the  $\alpha$ -helical content of a poly-Ala-based peptide measured  
739 from NMR for AMBER (AMBERff99SB and ff03) and CHARMM22 [140,141]. For developing  
740 an accurate FF for protein, it is critical to obtain the local conformational free energy surface of  
741 each amino acid residue without secondary structure constraint. The other important feature is  
742 the side-chain  $\chi^1$  rotamer distributions of residues other than Gly, Ala, and Pro. The FFs for  
743 carbohydrates are still under development due to their extended and dynamic structures. MD  
744 simulations were carried out on immunoglobulin-binding protein G (NuG2b) and other small  
745 proteins which showed that these occur between the native state and a highly collapsed, H-  
746 bonded species. However, it is reported that these small proteins show cooperativity during the  
747 folding process started from a denatured state that may be expanded and largely unstructured in  
748 the beginning even at the lower concentrations of denaturants. This excessive intramolecular H-  
749 bonding recorded in the simulations suggested that there is a need to change the FF parameters to  
750 achieve the balance between all the deriving players, H-bonding, backbone solvation, and non-  
751 bonded interactions [142]. The FFs for nucleic acids are developed to some extent and are still in  
752 evolving phase. The parameters for nucleic acids need be improved for refinement of torsion  
753 (sugar pucker and glycosidic torsion parameter), van der Waals parameters and for the point  
754 charges. CHARMM27 misjudges the groove asymmetry and probably base pairing stability in  
755 DNA (clearly this is the case in RNA) or underestimates twisted trajectories of B-DNA [143]. To  
756 obtain the conformational equilibrium for RNA is particularly challenging: it is single stranded  
757 and difficult to replicate the backbone substrates of many important motifs such as the sarcin-  
758 ricin loop, some tetraloops, tetranucleotides, reversed kink-turns, and ribozymes among others

759 [29]. MD analysis of lipid membranes has provided novel information about the functioning of  
760 biological systems such as lung surfactants and tear-fluid layer but still there is scope of  
761 improvement [17]. Therefore, the parameters of FFs need to be modified in such a manner so  
762 that they are not biased towards particular amino acid residues/secondary structure.

763

## 764 **8.2. Available force fields are deficient to study crowded environments of the cell**

765 The cell consists of different micro/macromolecules distributed in dynamically changing  
766 environment with different thermodynamic and kinetic properties of the proteins, carbohydrates,  
767 nucleic acids and other molecules. This intracellular accumulation of variety of biomolecules and  
768 inorganic molecules also affects the structural stabilities, intermolecular/intramolecular binding  
769 affinities and enzymatic rates of proteins. However, various structural biology methods are  
770 expensive in terms of resources and time, such as NMR or different spectroscopies, usually  
771 require samples with relatively high protein concentrations and purity. While, this is not the case  
772 *in vivo*, proteins are neither present in that high concentration nor exist in isolation but may  
773 remain crowded by numerous other biomolecules and ligands. The accuracy of classical MD  
774 simulations is not good enough to capture protein behaviour at high concentration as the current  
775 FFs parameters cannot describe the crowded microenvironment [144]. There is also need to  
776 develop parameters for water model, which seems to lack the strength required to quickly  
777 dissociate bad complexes or even prevent their formation [1]. However, recently Yu et al. (2016)  
778 have studied the protein-protein interactions, surface diffusion of secondary metabolites in two  
779 dimensional space and protein-ligand interactions in the bacterial cytoplasm by MD [145]. While  
780 this study has pointed towards the transferability and compatibility deficiency of the CHARMM  
781 FF for close molecular interactions under conditions where FF parameters have to be developed  
782 extensively and artefacts, e.g. over stabilization of protein-protein contacts, diffusion of  
783 secondary metabolites and inclusion of the cell membranes in the MD study. Therefore,  
784 scientific community working in this area needs to put substantial efforts to develop FFs with  
785 more specialized applications that can be helpful in studying the protein aggregates in a single  
786 simulation box or can analyze the effect of high ion concentration on protein. This should also  
787 include polarizable FFs that may be able to capture the redistribution of electrons around each  
788 atom in response to the dynamic changes in the environment around the molecules [131].

789

### 790 **8.3. Need to develop parameters for post-translational modifications**

791 There are several known post-translational modifications, phosphorylation being most common  
792 among them. The energetics of phosphorylated amino acids has been described by MD analysis  
793 which showed that the phosphorylated amino acids are involved in the hydrogen bond formation  
794 [146]. Since phosphate is a charged entity, it represents a crucial interacting site for molecular  
795 mechanics energy functions. The electronic polarizability cannot be treated by commonly used  
796 fixed charge FFs and is probably found as the significant feature for various portions of a  
797 biomolecule that may interact with a phosphate group. Yet, polarizable FFs can be used to  
798 analyze such interactions of biomolecules. However, post-translational modifications can  
799 modulate protein charge in the whole protein/specific manner and thus may also act as  
800 ‘electrostatic switches’. There are different types of post-translational modifications such as  
801 sulfonation, lysine acetylation, and glutamate methylation, the last two being important for  
802 histone modifications [84]. Therefore, there is a need to develop new and efficient parameters for  
803 van der Waals interactions which will occur between the atoms of protein and the covalently  
804 attached groups during PTM. PTMs are involved in bond stretching, angle bending, dihedral &  
805 improper torsions between the atoms of the protein and the attached group of atoms which may  
806 be consisted of carbon, phosphorous, sulphur and other atoms. However, the potential energy  
807 functions for the bond angle, dihedral torsion and electrostatic interactions are well developed for  
808 proteins. Nevertheless, proteins interact with various newly discovered chemical groups  
809 implicated by PTMs like acetyl, pyruvic acid, carboxyllysine etc., then, there is need to develop  
810 potential energy parameters for these changes which are not being explored using classical MD  
811 analysis and by experimental studies extensively yet.

812

### 813 **9. Conclusions and perspectives**

814 Recently, it has been reported that the current versions of popular FFs (including AMBER99SB-  
815 ILDN, CHARMM22-CMAP, GROMOS 45a3 and 54a7, and OPLS-AAL) are not accurate enough  
816 for studying the proteins existing in crowded environments because of the excessive tendency of  
817 interactions [144]. The events involved in the folding of the proteins in their native forms can be  
818 elucidated by MD simulations with molecular mechanics based FFs [147]. However, the current  
819 FFs have problems in describing the discrepancies in the local conformational preferences  
820 (secondary structure) of different amino acid residues [141]. Therefore, there is a serious need to

821 improve the existing FFs and also to develop additional parameters for different forms of  
822 proteins, nucleic acids and carbohydrates. There is also need for the development of polarisable  
823 FFs for studying the biomolecules occurring in crowded environments. However, there is an  
824 urgent need for developing easy protocols/methods so that the simulations can be carried out  
825 without much difficulty. Although, in case of lipid membranes, the FFs are well defined for  
826 various types of lipids but still there is a need to significantly improve the parameters for the  
827 protein-lipid (polar-head/tail) interactions, solvent-lipid interactions and solute-lipid interactions.  
828 There are many known MD reports that involve transition metals as co-factors of proteins that  
829 play important roles in the biological processes. Although, there are parameters developed for  
830 some of the monovalent and divalent metal ions in different FFs, however, there are still some  
831 known metals for which new parameters need to be developed. There is also a striving need of  
832 automation in the MD protocols so that, the researchers not having technical background in high  
833 end computing may also be able to use these protocols for carrying out the analysis for studying  
834 important biological phenomena. Easy going methods for MD usage protocols may change the  
835 prevailing scenario preferentially highly technical users to a popular tool for use by the common  
836 biologist folk. If all this is achieved in the near future, MD analysis definitively will be able to  
837 venture its way into the main stream of research in the basic and applied biology. There will be  
838 an integration of highly specialized “dry” labs with strong theoretical background and evident  
839 quantitative skills with the general biologists. Moreover, in this way, the tools and technologies  
840 developed in the “dry” labs will provide new opportunities resulting into leading methodologies  
841 which will be utilised in the “wet” labs, often exist without such backgrounds, but working in  
842 crucial aspects of the basic biology on different biomolecules.

843

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## 1284 **Figure legends**

### 1285 **Fig. 1. Comparative account of the conventional experimental biology techniques using** 1286 **molecular dynamics.**

1287 Before a biological experiment, the model organism needs to be first acclimatized with the lab  
1288 conditions. The molecular expression of gene, cloning and purification of the gene product and  
1289 the proteomic studies involve protein isolation, purification and enzyme assays to assess the  
1290 activity of the protein. The structure related studies are further complicated by the steps involved  
1291 in protein purification and crystallization. The MD analysis saves the time to be otherwise spent  
1292 for multiple random trials required for checking the efficiency of mutated proteins and for  
1293 understanding the possible mechanisms operating in such reactions.

1294

### 1295 **Fig. 2. Readily used different parameters need modifications in the current FFs.**

1296 Biomolecules are consists of same kind of atoms but the bond formation, bond angles and their  
1297 geometry and spatial distribution varies. (A) Different parameters are defined in the FFs in terms  
1298 of mathematical expression i.e. bond angle, bond length, bonded interactions and non-bonded  
1299 interactions in the amino acids of the protein. (B) Additional parameters for the torsion angle ( $\chi_1$ ,  
1300  $\chi_2, \chi_3, \chi_4$ ) of the side chain atoms that give rise to torsion parameters of the protein need to be  
1301 modified. The parameters for elucidating this aspect of the proteins are still under development.  
1302 (C) The parameters used for nucleic acids are defined on the bond angles of the backbone ( $\alpha$ ,  $\beta$ ,  
1303  $\gamma$ ,  $\delta$ ,  $\epsilon$ ,  $\zeta$ ). Since the structure of the nucleic acid is in extended form, however, the complete  
1304 parameters for explaining the same are not yet well defined. (D) Structure of carbohydrates is

1305 extended and dynamic in the solution. It is not well defined as that of the proteins, which is  
1306 defined by Ramachandran plot. FFs for carbohydrates are still need to be more developed.  
1307

1308 **Fig. 3. Flowchart for carrying out simple MD simulation experiments.**

1309 The quality of the biomolecular structure needs to be checked before starting the simulation  
1310 experiment. The functional unit of the biomolecule (monomer, dimer or oligomeric forms) needs  
1311 to be known. Then the input files are prepared and the biomolecule is immersed into the  
1312 simulation box. The biomolecule is energy minimized to obtain its most stable computed  
1313 conformation. The energy minimized biomolecule is equilibrated under constant temperature and  
1314 pressure conditions. Then it is simulated to study its structural behavior and conformational  
1315 changes.

1316

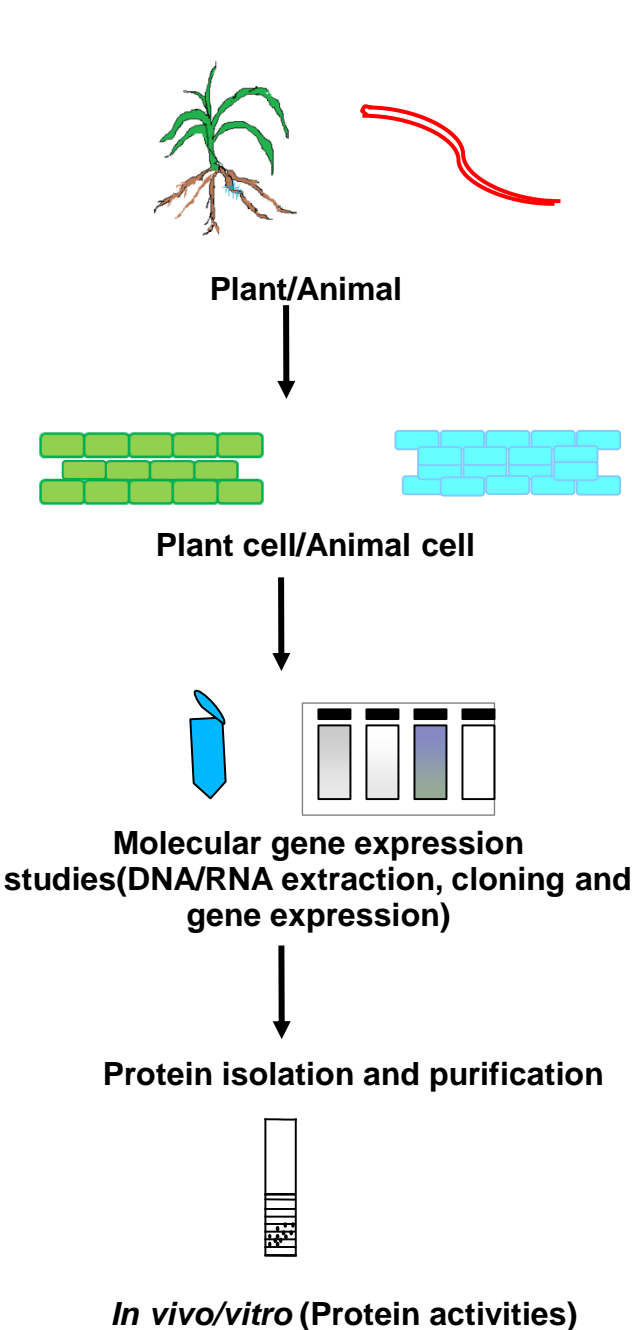
1317 **Fig. 4. Applications of molecular dynamics in biomolecular sciences.**

1318 MD approaches have been used to elucidate the structure, function and behaviour of  
1319 biomolecules in duplicate conditions in the simulation box. Different functions of the proteins  
1320 have been studied by MD analysis (catalysis, drug-binding locations, transport, motility, signal  
1321 transduction, allosteric regulation, protein-protein interaction specificity/binding, protein  
1322 embedded in lipids, protein folding/unfolding, post-translational changes, protein-secondary  
1323 small molecule interactions and molecular recognition of the proteins). The role of carbohydrates  
1324 involved in the signalling of various cellular processes, hydration of carbohydrates,  
1325 carbohydrate-protein (glycoprotein) interactions and carbohydrate-ligand interactions are  
1326 explained by MD. The conformational changes occurring at the molecular levels in DNA/RNA  
1327 have been successfully studied by MD.

1328

1329

**Fig.1.**



MHDLLQTWKADIDGKLLKYYQNSVDTNKSSLEKQIA  
KQKIFSKQDNIDLMVLNYYRVVQAYQSVHEKISLIF

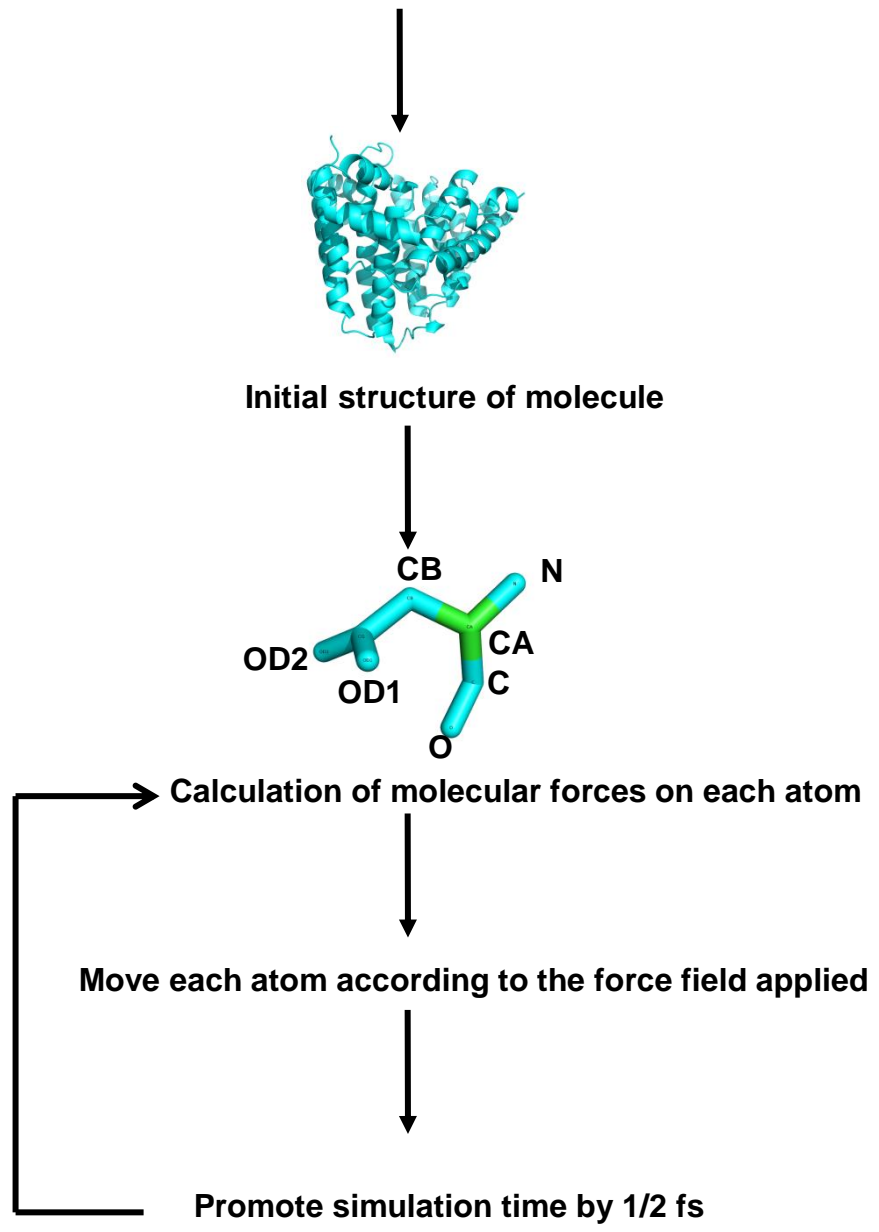


Fig.2.

